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Diagnosing CO₂ fluxes in the upwelling system off the Oregon–California coast

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Abstract. It is generally known that the interplay between the carbon and nutrients supplied from subsurface waters via biological metabolism determines the CO₂ fluxes in upwelling systems. However, quantificational assessment of such interplay is difficult because of the dynamic nature of both upwelling circulation and the associated biogeochemistry. We recently proposed a new framework, the Oceandominated Margin (OceMar), for semi-quantitatively diagnosing the CO₂ source/sink nature of an ocean margin over a given period of time, highlighting that the relative consumption between carbon and nutrients determines if carbon is in excess (i.e., CO₂ source) or in deficit (i.e., CO₂ sink) in the upper waters of ocean margins relative to their off-site inputs from the adjacent open ocean. In the present study, such a diagnostic approach based upon both couplings of physics-biogeochemistry and carbon-nutrients was applied to resolve the CO₂ fluxes in the well-known upwelling system off Oregon and northern California of the US west coast, using data collected along three cross-shelf transects from the inner shelf to the open basin in spring/early summer 2007. Through examining the biological consumption on top of the water mass mixing revealed by the total alkalinity-salinity relationship, we successfully predicted and semi-analytically resolved the CO₂ fluxes showing strong uptake from the atmosphere beyond the nearshore regions. This CO2 sink nature primarily resulted from the higher utilization of nutrients relative to dissolved inorganic carbon (DIC) based on their concurrent inputs from the depth. On the other hand, the biological responses to intensified upwelling were minor in nearshore waters off the Oregon–California coast, where significant CO_2 outgassing was observed during the sampling period and resolving CO_2 fluxes could be simplified without considering DIC/nutrient consumption, i.e., decoupling between upwelling and biological consumption. We reasoned that coupling physics and biogeochemistry in the OceMar model would assume a steady state with balanced DIC and nutrients via both physical transport and biological alterations in comparable timescales.

1 Introduction

The contemporary coastal ocean, characterized by high primary productivity due primarily to the abundant nutrient inputs from both river plumes and coastal upwelling, is generally seen as a significant CO₂ sink at the global scale (Borges et al., 2005; Cai et al., 2006; Chen and Borges, 2009; Laruelle et al., 2010; Borges, 2011; Cai, 2011; Dai et al., 2013). However, mechanistic understanding of the coastal ocean carbon cycle remains limited, leading to the unanswered question of why some coastal systems are sources while others are sinks of atmospheric CO₂ in a given timescale. We recently proposed a new framework, the Ocean-dominated Margin (Oce-Mar), for better shaping the concept of a coastal ocean carbon study (Dai et al., 2013). This framework highlights the importance of the boundary process between the open ocean and the ocean margin, and proposes a semi-analytical diagnostic approach to resolve sea-air CO₂ fluxes over a given period of time. The approach invokes an establishment of the water mass mixing scheme in order to define the physical transport of, or the conservative portion of carbon and nutrients, from the adjacent open ocean, and the constraint of the biogeochemical alteration of these non-local inputs in the upper waters of ocean margins. The water mass mixing scheme is typically revealed using conservative chemical tracers such as total alkalinity (TAlk) and/or dissolved calcium ions (Ca^{2+}) to bypass the identification of end members associated with individual water masses that often possess high complexity in any given oceanic regime. The constraint of the biogeochemical alteration can then be estimated as the difference between the predicted values based on conservative mixing between end members and the field-measured values. The relative consumption between dissolved inorganic carbon (DIC) and nutrients determines if DIC is in excess or in deficit relative to the off-site input. Such excess DIC will eventually be released to the atmosphere through sea-air CO2 exchange. Using two large marginal seas, the South China Sea (SCS) and the Caribbean Sea (CS), as examples, we have successfully predicted, via evaluating DIC and nutrient mass balance, the CO₂ outgassing that is consistent with field observations (Dai et al., 2013). However, the OceMar concept and the diagnostic approach have not been verified on upwelling systems that can be either sources (e.g., Friederich et al., 2002; Torres et al., 2003; Fransson et al., 2006) or sinks (e.g., Borges and Frankignoulle, 2002; Santana-Casiano et al., 2009; Evans et al., 2012) of atmospheric CO₂. While it is generally known that the interplay between the nutrients and DIC supplied from subsurface waters via biological metabolism would determine the CO₂ fluxes in upwelling systems, quantificational assessment of such interplay is difficult because of the dynamic nature of both upwelling circulation and the associated biogeochemistry.

Our study therefore chose the upwelling system offshore Oregon and northern California, for examining the CO₂ flux dynamics during the upwelling season through our proposed mass balance approach associated with carbon/nutrient coupling. The system under study is part of the eastern boundary current in the North Pacific (Fig. 1). While strong equatorward winds in spring/summer drive offshore Ekman transport at the surface over the coastal waters, the carbon- and nutrient-rich deep water is transported shoreward and upward over the shelf to compensate for the offshore transport in the surface layer (Huyer, 1983; Kosro et al., 1991; Allen et al., 1995; Federiuk and Allen, 1995; Gan and Allen, 2002). Outcrops of waters from depths of 150-200 m are frequently observed nearshore the Oregon-California shelf, where the surface partial pressure of CO_2 (pCO_2) can reach levels near 1000 µatm. This water is then transported seaward and south-

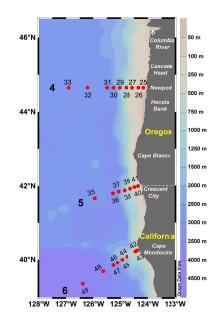


Figure 1. Map of offshore Oregon and northern California (US west coast) showing the topography and the locations of sampling stations along transects 4, 5 and 6 in spring/early summer 2007.

ward while the pCO_2 is drawn down by biological productivity, and can reach values down to ~ 200 µatm, far below the atmospheric pCO_2 value (Hales et al., 2005, 2012; Feely et al., 2008; Evans et al., 2011). Such a dramatic decrease in seawater pCO_2 may be due to the fact that the complete utilization of the preformed nutrients in upwelled waters exceeds their corresponding net DIC consumption, leading to the area off Oregon and northern California acting as a net sink of atmospheric CO₂ during the upwelling season (Hales et al., 2005, 2012). On the other hand, Evans et al. (2011) suggest that the spring/early summer undersaturated pCO_2 conditions in some offshore areas result from non-local productivity associated with the Columbia River (CR) plume, which transports ~77 % of the total runoff from western North America to the Pacific Ocean (Hickey, 1989).

In this context, the Oregon–California shelf in the upwelling season could be a potential OceMar-type system, with the majority of DIC and nutrients in the upper layer originating from the non-local deep waters in the subtropical gyre of the eastern North Pacific (eNP), though riverine inputs might complicate the application of the OceMar framework. On the other hand, the upper waters in offshore areas beyond the upwelling circulation on the Oregon–California shelf would be largely fed by on-site deep waters via vertical mixing, with minor influence of the CR plume.

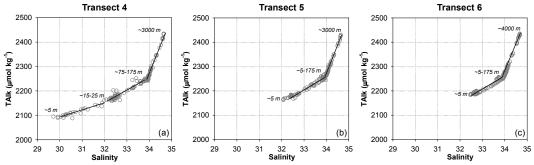


Figure 2. Total alkalinity versus salinity throughout the entire water column of sampling stations along transects 4 (a), 5 (b) and 6 (c) off Oregon and northern California in spring/early summer 2007. The solid lines indicate various linear relationships observed on each transect. The numbers in italics denote the sampling depth/depth range of the endpoints of each line.

2 Study area and data source

2.1 California Current system and upwelling circulation

The upwelling circulation off Oregon and northern California is linked with the eastern boundary current, the California Current (CC), occupying the open basin of the eNP (Barth et al., 2000). The CC is a broad and weak surface current (0–200 m) which carries low-salinity/low-temperature water equatorward from the sub-Arctic Pacific (Lynn and Simpson, 1987). The deeper-lying California Undercurrent (CUC, 150–300 m), which has relatively high salinity and temperature, originates in the eastern equatorial Pacific and flows poleward inshore along the west coast of North America (Thomson and Krassovski, 2010). The CC system is characterized by coastal upwelling in spring/summer, during which waters primarily composed of the CC are transported upward from the depths of 150–200 m towards the nearshore surface off the Oregon–California coast (Castro et al., 2001).

Both field observations and modeling studies (Oke et al., 2002; Gan and Allen, 2005) show that the upwelling circulation pattern in the study area differs significantly between north and south of Newport (Fig. 1). North of Newport between 45.0 and 45.5° N with a relatively straight coastline and narrow shelf, the alongshore uniform bottom topography generally results in typical upwelling circulation with a southward coastal jet close to shore at Cascade Head (Fig. 1). Over the central Oregon shelf between 43.5 and 45.0° N, the highly variable bottom topography over Heceta Bank (Fig. 1) largely influences the upwelling circulation, leading to a complex three-dimensional flow pattern with offshore shifting of the coastal jet and development of northward flow inshore. At the coast along the southern part of Oregon and northern California between 39.0 and 43.0° N, an enhancement of coastal upwelling, jet separation and eddy formation are observed to be associated with interactions of the windforced coastal currents with Cape Blanco (Fig. 1) (Barth and Smith, 1998; Gan and Allen, 2005, and references therein).

2.2 Data source

Our data sets were based on the online-published carbonate system and nutrient data collected along three transects off Oregon and northern California during the first North American Carbon Program (NACP) West Coast Cruise in spring/early summer (11 May–14 June) 2007 (http://cdiac.ornl.gov/oceans/Coastal/NACP_West.html; Feely et al., 2008; Feely and Sabine, 2011). Transect 4 (stations 25–33 from nearshore to offshore) is located off Newport, Oregon. Transect 5 (stations 41–35 from nearshore to offshore) is located off Crescent City near the Oregon–California border. Transects 6 (stations 42–49 from nearshore to offshore) is located off Cape Mendocino, California. The majority of the offshore stations on all transects were located in the open subtropical gyre of the eNP (Fig. 1).

3 Results and discussion

The region under study is highly dynamic, potentially involving coastal upwelling, the CR plume and pelagic waters mixed by various Pacific water masses (Hill and Wheeler, 2002). Instead of accounting for all of the water masses contributing to the CC system, the mixing scheme in the upper waters along the three transects was examined via the total alkalinity-salinity (TAlk-Sal) relationship obtained during the sampling period so as to quantify the conservative portion of DIC and nitrate (NO₃). The end members were therefore identified under this relationship, which might have experienced physical or biological alterations from their original water masses such as the CR and the CC. Subsequently, the biologically consumed DIC and NO₃ were quantified as the difference between their conservative values predicted from the derived end-member mixing and the corresponding field measurements. Finally, the CO₂ source/sink nature of the upper waters off Oregon and northern California during the sampling period was diagnosed via a mass balance approach by estimating the relative consumption between DIC

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and NO_3 , and a simple sensitivity analysis was performed to test the robustness of the approach.

3.1 TAlk–Sal relationship

3.1.1 Throughout the entire water column off Oregon and northern California

Three generally linear relationships between TAlk and salinity were observed throughout the entire water column along transect 4 (Fig. 2a). The first one was for waters with salinity lower than ~ 32.0 (corresponding to a depth of $\sim 15-25$ m), which were significantly influenced by the CR plume. The second one was for waters composed primarily of the CC with salinity between ~ 32.0 and ~ 33.9 , including those immediately below the surface buoyant layer at stations 26-32 and the surface waters at the outermost station 33 (Fig. 1). The higher-end salinity value of ~ 33.9 corresponded to a depth range of \sim 75–175 m, composed possibly of the upwelled high-salinity CUC waters. At station 27 (water depth \sim 170 m) for instance, salinity at depths of \sim 130 and \sim 160 m reached \sim 34.0 with TAlk values of \sim 2260 µmol kg⁻¹, which were even higher than those of offshore waters at $\sim 175 \text{ m}$ ($\sim 2250 \text{ µmol kg}^{-1}$). These two data points were thus located on the third linear relationship for waters with salinity higher than \sim 33.9, the slope of which became much steeper, mainly reflecting the mixing between the approaching CUC and deep waters in the subtropical gyre of the eNP (Fig. 2a).

All salinity values, including surface samples on transects 5 and 6, were higher than 32.0 (Fig. 2b and c). With minor influence of the CR plume, the TAlk–Sal relationship displayed two generally linear phases throughout the entire water column along both transects, while the TAlk and salinity endpoints of each were comparable to those of the latter two observed on transect 4 (i.e., the linear phases for waters with salinity between \sim 32.0 and \sim 33.9 and with salinity higher than \sim 33.9; Fig. 2a). Note that the turning point with salinity of \sim 33.9 corresponded to a wider depth range of \sim 5–175 m (Fig. 2b and c), resulting from the most intensive upwelling on transects 5 and 6 bringing deep waters to the nearshore surface (Feely et al., 2008).

As suggested by the generally linear TAlk–Sal relationships, surface waters beyond the CR plume and waters immediately below the surface buoyant layer were directly linked to the underlying waters to the depth of ~ 175 m. We thus took a closer look at the TAlk–Sal relationship in the upper 175 m waters off Oregon and northern California.

3.1.2 In the upper 175 m waters off Oregon and northern California

In the upper 175 m waters along transect 4, the linear regression for waters with salinity lower than ~ 32.0 had an intercept of $\sim 1200 \,\mu\text{mol kg}^{-1}$. This value agreed well with

the observed TAlk of $\sim 1000 \,\mu\text{mol}\,\text{kg}^{-1}$ in the mainstream of the CR (Park et al., 1969b; Evans et al., 2013). The other linear regression for waters with salinity between \sim 32.0 and \sim 33.9 had a smaller intercept of \sim 500 µmol kg⁻¹, implying a smaller contribution from the CR plume (Fig. 3a). Exceptions were observed at the shallowest station 25 (water depth \sim 50 m) and the deepest station 33 (water depth \sim 2900 m). The TAlk-Sal relationship completely followed the second phase for the upper 175 m waters at station 33 (Fig. 3a), suggesting a small fraction of the CR plume even in the surface waters of this outermost station on transect 4. On the other hand, data points of the two variables were not well correlated throughout the entire water column of station 25 and fell off both regression lines (Fig. 3a). The water mass mixing at this innermost station was not as straightforward, despite minor freshwater admixture as suggested by the high surface salinity of > 32.0.

The TAlk–Sal relationship in the upper 175 m waters on transects 5 and 6 displayed two similar phases. One was the linear regression for stations 35–38 (deeper than $\sim 800 \,\mathrm{m}$) and stations 45–49 (deeper than \sim 1400 m), with slope and intercept values comparable to the second phase observed on transect 4. The other was the linear regression for the three shallow stations on both transects largely influenced by coastal upwelling (Feely et al., 2008) (Fig. 3b and c). This phase was not clearly seen from the full TAlk-Sal plot (Fig. 2b and c), as the salinity in the upper 175 m waters at stations 39 and 44 as well as in the entire water column of stations 40-43 varied within a much smaller range of \sim 33.3– \sim 34.0. The negligible intercepts of this TAlk–Sal regression suggested insignificant freshwater input with zero solutes to the intensive upwelling zone off Oregon and northern California (Fig. 3b and c).

All phases shown in Fig. 3 displayed good linear TAlk–Sal relationships (r > 0.94), indicating an overall twoend-member mixing scheme for each phase. Although the non-conservativity of TAlk existed, it was not that significant as seen by the deviations of a few data points from each linear regression (Fig. 3). As a matter of fact, Fassbender et al. (2011) estimated that the contribution from CaCO₃ dissolution to the TAlk addition in the surface mixed layer on transect 5 was $< 10 \,\mu\text{mol}\,\text{kg}^{-1}$ (< 0.5 % of their absolute contents in seawater) and well near the analytical precision. Such small non-conservative portions would not compromise the application of TAlk as a conservative tracer. Note that the two-end-member mixing was not spatially homogeneous in the upper waters off Oregon and northern California during the sampling period. The surface waters at stations 26-32 on transect 4 were imprinted by the CR plume with a salinity around \sim 30.0. During the transport from the mouth of the CR estuary, the plume water increasingly mixed with adjacent oceanic waters, largely feeding its pathway. However, the majority of DIC and nutrients in waters immediately below the buoyant layer, as well as in surface waters at station 33 and possibly at station 25, originated from deep

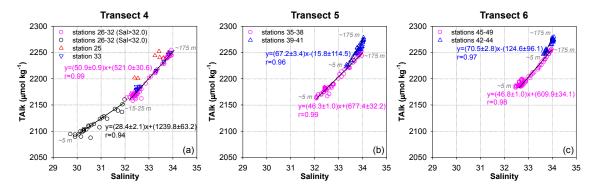


Figure 3. Total alkalinity versus salinity (TAlk–Sal relationship) in the upper 175 m waters of sampling stations along transects 4 (**a**), 5 (**b**) and 6 (**c**) off Oregon and northern California in spring/early summer 2007. The solid lines as well as the equations (in accordance with the symbol colors) indicate the linear regression analyses of the TAlk–Sal relationship for various stations. The numbers in italics denote the sampling depth/depth range of the endpoints of each line. In (**a**), the TAlk–Sal relationship at station 26–32 displayed two phases for waters with salinities lower and higher than \sim 32.0. The surface waters at these stations were imprinted by the Columbia River plume. The data points of bottom waters at stations 26 (\sim 75 m) and 27 (\sim 130 and \sim 160 m) were not included, as they were located on the third linear relationship shown in Fig. 2a. In (**b**) and (**c**), stations 39–41 and stations 42–44 were largely influenced by coastal upwelling.

waters, brought up through coastal upwelling and/or vertical mixing. The influence of the CR plume still occurred but was diluted by other freshwater masses such as rainwater, suggesting a mixing scheme between the deep water in the subtropical gyre of the eNP and a combined freshwater end member (Park, 1966, 1968). Such mixing was also applicable to the surface waters at stations 35–38 on transect 5 and stations 45–49 on transect 6. On the other hand, the upper 175 m waters or the entire water column at stations 39–44 resulted from a simple two-end-member mixing between the upwelling source water and the rainwater with zero solutes, pointing to an apparent OceMar-type system.

3.2 \triangle DIC and \triangle NO₃ in the upper waters off Oregon and northern California

The defined mixing schemes enabled us to estimate the nonconservative portion of DIC (Δ DIC) and NO₃ (Δ NO₃) in the upper waters off Oregon and northern California following Dai et al. (2013):

$$\Delta \text{DIC} = \text{DIC}^{\text{cons}} - \text{DIC}^{\text{meas}},\tag{1}$$

$$\Delta NO_3 = NO_3^{cons} - NO_3^{meas}, \qquad (2)$$

$$X^{\text{cons}} = \frac{\text{Sal}^{\text{meas}}}{\text{Sal}^{\text{ref}}} \cdot (X^{\text{ref}} - X^{\text{eff}}) + X^{\text{eff}}.$$
 (3)

The superscripts "cons" and "meas" in Eqs. (1) and (2) denote conservative-mixing-induced and field-measured values. In Eq. (3), X represents DIC or NO₃, while Sal^{meas} is the Conductivity–Temperature–Depth (CTD) recorder-measured salinity. Sal^{ref} and X^{ref} are the reference salinity

and concentration of DIC or NO₃ for the deep water end member, which are the averages of all ~ 175 m samples from stations involved in each mixing scheme. Specifically, for waters immediately below the surface buoyant layer at stations 27–32 and waters in the surface mixed layer at stations 25 and 33 on transect 4, the deep water end-member values of the reference salinity and concentrations of DIC or NO₃ were the averages of ~ 175 m samples from stations 28–33 (Fig. 1). On transects 5 and 6, the preformed salinity, DIC and NO₃ values for waters in the surface mixed layer at stations 35–38 and at stations 45–49 were the averages of ~ 175 m samples of these stations. For the upper waters influenced by the intensified upwelling at stations 39–41 and stations 42–44, the deep water end member was selected as the ~ 175 m water at station 39 and at station 44 (Fig. 1).

The X^{eff} in Eq. (3) denotes the effective concentration of DIC or NO₃ sourced from the freshwater input to various zones off Oregon and northern California. Since rainwater was assumed to have no solutes, both DIC^{eff} and NO₃^{eff} would be zero for waters in the surface mixed layer of stations 39–41 on transect 5 and stations 42–44 on transect 6. On the other hand, the estimation of X^{eff} associated with the CR followed the method for the OceMar case study of the CS, which has a noticeable DIC^{eff} from the combination of the Amazon River and the Orinoco River (Dai et al., 2013).

Since bicarbonate dominates other CO₂ species and other alkalinity components, DIC concentrations in the mainstream of the CR are numerically similar to TAlk, which are also around $\sim 1000 \,\mu\text{mol}\,\text{kg}^{-1}$ (Park et al., 1969a, 1970). This value was taken as the DIC end member of the CR. The NO₃ end-member value was selected as 15 μ mol kg⁻¹ based on recent years' observations in May and June at station SATURN-05 established in the upstream CR (database of the Center for Coastal Margin Observation and Prediction; http://www.stccmop.org/datamart/). Assuming that the biological consumption of DIC and NO₃ in the CR plume followed the Redfield ratio (Redfield et al., 1963), the DIC removal was estimated to be ~ 100 µmol kg⁻¹ (approximately 15 × 106/16), while NO₃ was rapidly consumed along the pathway of the CR plume and generally depleted in the area beyond the plume (Aguilar-Islas and Bruland, 2006; Lohan and Bruland, 2006). As a consequence, the complete DIC^{eff} and NO^{eff}₃ in the upper waters from the CR would be ~ 900 and ~ 0 µmol kg⁻¹.

If the combined freshwater end member was a mixture of the CR and the rainwater with zero solutes, the intercept values of 521.0 ± 30.6 (Fig. 3a), 677.4 ± 32.2 (Fig. 3b) and 609.9 ± 34.1 (Fig. 3c) derived from the TAlk-Sal regression would indicate that the CR fractions were ~ 50 , ~ 65 and $\sim 60\%$ (approximately 500/1000, 650/1000 and 600/1000 taking ~ 1000 µmol kg⁻¹ as the TAlk end-member value of the CR, Park et al., 1969b; Evans et al., 2013). The DIC^{eff} from the freshwater input was thus estimated to be $\sim 450 \,\mu\text{mol}\,\text{kg}^{-1}$ (approximately $900 \times 50 \,\%$) for waters immediately below the surface buoyant layer at stations 27-32 and waters in the surface mixed layer at stations 25 and 33 on transect 4, which was slightly lower than the $\sim 585\,\mu mol\,kg^{-1}$ (approximately $900\times 65\,\%)$ and the $\sim 540 \,\mu\text{mol}\,\text{kg}^{-1}$ (approximately 900 \times 60 %) for waters in the surface mixed layer at stations 35-38 on transect 5 and at stations 45–49 on transect 6, respectively. The NO_2^{eff} in any combined freshwater end member was zero.

Note that numerous small mountain rivers are distributed along the Oregon-California coast, which might also dilute the CR plume, inducing the lower intercept of the TAlk-Sal regression observed on the three transects (Fig. 3). The average wintertime discharge from the Oregon Coast Range rivers is estimated to be $\sim 2570 \text{ m}^3 \text{ s}^{-1}$ (Wetz et al., 2006), which is more than an order of magnitude higher than that in the summer (Colbert and McManus, 2003; Sigleo and Frick, 2003). However, the CR discharge in May to June 2007 reached its maximum of $\sim 15000 \text{ m}^3 \text{ s}^{-1}$ (Evans et al., 2013), which should be approximately 2 orders of magnitude higher than the discharge of small rivers. This significant contrast would suggest that inputs from small rivers should be negligible compared to the CR plume. In particular, inputs from small rivers are normally restricted to a narrow band near the coast, whereas the research domain of this study extended to the open subtropical gyre of the eNP. Even the surface salinity at the innermost stations (i.e., station 25 on transect 4, station 41 on transect 5 and station 42 on transect 6; Fig. 1) was as high as \sim 32.5, \sim 33.9 and \sim 34.0, which would rule out the influence of small rivers.

3.3 Evaluating the CO₂ source/sink nature in the upper waters off Oregon and northern California

The coupling of DIC and NO₃ dynamics could then be examined based on the classic Redfield ratio of C: N = 106: 16 = 6.6 (Redfield et al., 1963). Positive values of the difference between Δ DIC and 6.6 Δ NO₃ (Δ DIC-6.6 Δ NO₃) suggested a CO₂ source term since "excess Δ DIC" was removed by CO₂ degassing into the atmosphere. In contrast, negative Δ DIC-6.6 Δ NO₃ suggested that "deficient Δ DIC" was supplied via the atmospheric CO₂ input to the ocean, representing a CO₂ sink. Such net CO₂ exchange between the seawater and the atmosphere was further quantified as the sea-air difference of *p*CO₂ (Δ *p*CO₂) via the Revelle factor (RF), which is referred to as the fractional change in seawater CO₂ over that of DIC at a given temperature, salinity and alkalinity, and indicates the ocean's sensitivity to an increase in atmospheric CO₂ (Revelle and Suess, 1957; Sundquist et al., 1979). Because *p*CO₂ and CO₂ are proportional to each other, the RF can be illustrated as:

$$RF = \frac{\partial p CO_2 / p CO_2}{\partial DIC / DIC}.$$
(4)

Here, $\partial p CO_2$ and ∂DIC are the fractional changes of pCO_2 and DIC in the surface seawater. In a simplified way and as an approximation, ∂DIC equals $\Delta DIC - 6.6 \Delta NO_3$, which is solely achieved through sea-air CO₂ exchange in the Oce-Mar framework, implying that ∂pCO_2 may represent the sea-air ΔpCO_2 . Given an initial balance of CO₂ between the seawater and the atmosphere, the sea-air ΔpCO_2 is obtained by

Sea-air
$$\Delta p CO_2 = \partial p CO_2 = RF \cdot p CO_2 \cdot \frac{\partial DIC}{DIC}$$
 (5)
= $RF \cdot p CO_2^{air} \cdot \frac{\Delta DIC - 6.6 \Delta NO_3}{DIC}$.

As shown in Fig. 4, the estimated $\Delta DIC-6.6\Delta NO_3$ values and their corresponding sea–air ΔpCO_2 values in the upper waters off Oregon and northern California were overall below zero, suggesting a significant CO₂ sink nature in the upwelling season.

3.3.1 Transect 4

On transect 4 off Newport, the average value of $\Delta DIC - 6.6 \Delta NO_3$ was $-23 \pm 2 \,\mu mol \, kg^{-1}$ in waters immediately below the surface buoyant layer at stations 27-32, which equaled the average value for the surface mixed layer at station 33 (Fig. 4a). Note that we were not able to derive values of $\Delta DIC - 6.6 \Delta NO_3$ at station 26 where NO₃ data were not available. Although located at different depths, the two water parcels experienced similar physical mixing and biogeochemical modifications inducing the same CO_2 signature. The former water mass should work as a CO₂ sink when in contact with the atmosphere before or after the passage of the episodic CR plume. The average sea-air $\Delta p CO_2$ resulting from the combined deficient ΔDIC was $-54 \pm 4 \mu$ atm (Fig. 3a). Given the atmospheric pCO₂ of \sim 390 µatm (Evans et al., 2011), the seawater pCO₂ in these regions was thus estimated to be $336 \pm 4 \mu atm$, which agreed

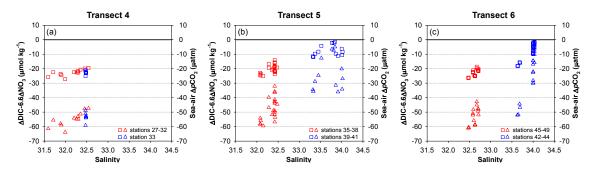


Figure 4. Δ DIC-6.6 Δ NO₃ (squares) and sea-air Δ *p*CO₂ (triangles) versus salinity in the upper waters on transects 4 (a), 5 (b) and 6 (c) off Oregon and northern California in spring/early summer 2007. Note that data for stations 27–32 on transect 4 were obtained from waters immediately below the surface buoyant layer, while data for other stations were obtained from the surface mixed layer. The value of 6.6 is the Redfield C/N uptake ratio (approximately 106/16; Redfield et al., 1963). The solid line indicates the *p*CO₂ equilibrium between the seawater and the atmosphere.

rather well with the field measurements of $334 \pm 13 \mu atm$ (the underway seawater pCO_2 data were not available online but alternatively calculated by applying TAlk and DIC data to the CO2SYS program; Lewis and Wallace, 1998).

The diagnostic approach was not applied to the surface buoyant layer since the aged CR plume might have experienced complex mixing with various surrounding water masses during its transport, as indicated by the scatter TAlk-Sal relationship (Fig. 3a). However, the far-field CR plume is suggested to be a strong sink of atmospheric CO₂ due to earlier biological consumption (Evans et al., 2011), which was supported by the observed low pCO_2 of \sim 220–300 µatm in the surface buoyant layer on transect 4. As a consequence, the CO_2 sink nature in the upper waters from the outer shelf (the bottom depth of station 27 was \sim 170 m) to the open basin off Newport, Oregon would primarily result from the higher utilization of nutrients relative to DIC based on their concurrent inputs from deep waters. The non-local high productivity in the CR plume could inject even lower pCO_2 , but this effect would be transitory.

At the innermost station 25 on transect 4, highly positive values of $\Delta DIC - 6.6 \Delta NO_3$ and sea-air ΔpCO_2 $(\sim 82 \,\mu mol \, kg^{-1}$ and $\sim 157 \,\mu atm$, respectively) were obtained for the surface mixed layer of this station, indicating a significant CO_2 source. However, the lowest pCO_2 value of \sim 170 µatm was observed in these nearshore waters off Oregon. The poor correlation between TAlk and salinity at station 25 (Fig. 3a) might compromise the estimation, whereas the same method (Eqs. 1-5) was successfully applied to other stations on transect 4 with a distinct TAlk-Sal relationship (i.e., the second phase in Fig. 3a). Note that coastal upwelling clearly influenced the bottom water at station 25 as indicated by the comparable salinity and TAlk values to those in offshore 200 m waters. Instead of being fed by the upwelled deep water, the DIC and nutrients in the surface mixed layer might have originated from horizontal admixture of the surrounding waters. These waters possibly experienced intense diatom blooms due to the fact that the surface silicate concentrations at station 25 were almost zero, which led to the most undersaturated pCO_2 condition observed in the upper waters off Oregon.

3.3.2 Transects 5 and 6

On transect 5 near the Oregon–California border, the average $\Delta DIC-6.6\Delta NO_3$ and sea–air ΔpCO_2 were estimated to be $-20 \pm 3 \,\mu$ mol kg⁻¹ and $-48 \pm 8 \,\mu$ atm in the surface mixed layer of stations 35–38 (Fig. 4b). Both values were comparable to those obtained from the surface mixed layer of stations 45–49 on transect 6 ($-23 \pm 3 \,\mu$ mol kg⁻¹ and $-53 \pm 6 \,\mu$ atm, respectively; Fig. 4c) and on transect 4, indicating a similar magnitude of the CO₂ sink term in offshore areas along the Oregon and northern California coast during the sampling period. The estimated sea surface pCO_2 of $342 \pm 8 \,\mu$ atm for transect 5 and $337 \pm 6 \,\mu$ atm for transect 6 were consistent with the field measurements of 332 ± 12 and $346 \pm 12 \,\mu$ atm in these regions.

The diagnosed CO₂ flux in the nearshore was also comparable between transects 5 and 6. The Δ DIC-6.6 Δ NO₃ and sea-air Δp CO₂ in the surface mixed layer of stations 39–44, although still below zero, were obviously higher than those of stations 35–38 on transect 5 and of stations 45–49 on transect 6 (Fig. 4b and c). Such an increase was expected since stations 39–44 were located in the area with the most intensive upwelling, which brought CO₂-rich deep waters to the nearshore surface (Feely et al., 2008). However, our estimation suggested a weaker CO₂ sink or close to being in equilibrium with the combined estimated sea surface pCO₂ of 368 ± 14 µatm, whereas the field measurements of ~ 600–1000 µatm indicated that the coastal upwelling zone should be a very strong source of CO₂ to the atmosphere.

Therefore, we took a closer look at transect 5: a uniform salinity of ~ 34.0 throughout the entire water column was observed at stations 40 and 41 due to the outcrop of the upwelling source water at the surface of the inner shelf

on transect 5 (Feely et al., 2008). Although salinity in the surface mixed layer at station 39 was lower, around \sim 33.4, the dilution effect of rainwater should be negligible. After removing the rainwater from the mixing scheme and calculating Δ DIC and Δ NO₃ by directly subtracting the fieldobserved value from the end-member value for the upwelling source water (Eqs. 6 and 7; DICref and NO3ref were field measurements of $\sim 200 \,\mathrm{m}$ water samples at station 39), the $\Delta DIC - 6.6 \Delta NO_3$ values were rapidly increased to above zero in the surface mixed layer at stations 39 and 40, while values at station 41 with a small increase were still overall below zero (Fig. 5a). Correspondingly, the estimated sea surface pCO_2 values were higher than the atmospheric CO_2 value at stations 39 and 40 while they were slightly lower than that at station 41. However, these values still largely fell below the field measurements of seawater pCO_2 , displaying shoreward increasing differences from ~ 200 to $\sim 700 \,\mu atm$ (Fig. 5b).

$$\Delta \text{DIC} = \text{DIC}^{\text{ref}} - \text{DIC}^{\text{meas}},\tag{6}$$

$$\Delta NO_3 = NO_3^{\text{ref}} - NO_3^{\text{meas}}.$$
 (7)

With or without taking rainwater into account, our diagnostic approach did not work in the nearshore areas with strong upwelling off Oregon and northern California, even though the mixing scheme of this region was in accordance with the OceMar concept. We contend that OceMar assumes a steady state with balanced DIC and nutrients via both physical mixing and biological alterations in comparable timescales. However, the continuous inputs from the coastal upwelling might have led to the accumulation of DIC and nutrients in the nearshore surface, which could not be timely consumed by the phytoplankton community, suggesting a possible nonsteady state. Fassbender et al. (2011) estimated that the age of the surface mixed layer at nearshore stations on transect 5 is only ~ 0.2 days, during which the DIC and NO₃ consumption via organic carbon production was almost zero, and CaCO₃ dissolution contributed a small fraction to the slightly elevated DIC in the upwelled waters. They further predicted that the nearshore surface pCO_2 on transect 5 will decrease to levels of $\sim 200 \,\mu atm$ in ~ 30 days until NO₃ exhaustion via continued biological productivity, implying the achievement of a steady state (Fassbender et al., 2011). Minor biological responses during the intensified upwelling period were also observed in summer 2008, allowing highly oversaturated pCO_2 surface water to persist on the inner shelf off Oregon for nearly 2 months (Evans et al., 2011). At this point, it is uncertain why there was such a prolonged delay from the phytoplankton community to the persistent source of upwelled DIC and nutrients. Note that under the condition of a more prevailing upwelling-favorable wind as a predicted consequence of climate change (e.g., Snyder et al., 2003; Diffenbaugh et al., 2004; Sydeman et al., 2014), the nearshore

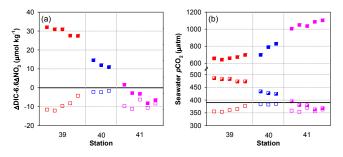


Figure 5. Panel (a) shows $\triangle DIC - 6.6 \triangle NO_3$ and (b) shows seawater pCO_2 in the surface mixed layer at stations 39–41 on transect 5 near the Oregon-California border in spring/early summer 2007. In (a), open symbols indicate values estimated based on the twoend-member mixing between the upwelling source water and the rainwater, while filled symbols indicate values after removing the rainwater. The value of 6.6 is the Redfield C/N uptake ratio (approximately 106/16; Redfield et al., 1963). The solid line indicates the pCO_2 equilibrium between the seawater and the atmosphere. In (b), the open and semi-filled symbols denote the estimated sea surface pCO_2 from $\Delta DIC - 6.6 \Delta NO_3$ on top of the mixing with and without rainwater, respectively. The filled symbols denote the field-observed sea surface pCO_2 , which were obtained by applying TAlk and DIC data into the CO2SYS program (Lewis and Wallace, 1998). The solid line denotes the atmospheric pCO_2 of ~ 390 µatm (Evans et al., 2011).

waters off the Oregon–California coast in the upwelling season might always be in a non-steady state, and it is expected that fewer periodic relaxation events or reversals would further decrease the chance for the biological response to be factored in.

In addition, the negligible biological consumption might involve large errors when calculating Δ . The portion of Δ DIC and Δ NO₃ at station 41 relative to the preformed values of the upwelling source water were only ~ 0.5 and ~ 10 %, slightly higher than the measurement uncertainties. The portion of DIC and NO₃ consumption in the surface mixed layer at offshore stations on transect 5 were, however, 1 order of magnitude higher (\sim 7 and \sim 90%, respectively). This contrast might partially explain why the Oce-Mar framework did not work when insignificant biological alterations occurred. Given the predominant control of physical mixing, we contend that the prediction of the CO_2 flux in the nearshore off Oregon and northern California with intensified upwelling could be simplified without considering DIC/nutrient consumption. In other words, surface CO2 levels in this region were simply imprints of the upwelling source water ($pCO_2 \sim 1000 \,\mu atm$ at $\sim 150-200 \,m$) with minor dilution by rainwater.

3.4 Sensitivity analysis

In the above exercise, both the end-member values revealed in the water mass mixing scheme and the Redfield ratio adopted in coupling DIC and nutrients were critically

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Table 1. Values of $\Delta DIC - 6.6 \Delta NO_3$, sea-air ΔpCO_2 and sea surface pCO_2 estimated with different DIC ^{eff} , which is the combined fresh-
water end member of DIC partly sourced from the Columbia River (CR).

$\begin{array}{l} TAlk/DIC \ of \ CR \\ (\mu mol \ kg^{-1}) \end{array}$	$NO_3 ext{ of } CR$ ($\mu mol ext{ kg}^{-1}$)	DIC ^{eff} (µmol kg ⁻¹)	$\frac{\Delta \text{DIC}-6.6\Delta \text{NO}_3}{(\mu \text{mol } \text{kg}^{-1})}$ Transect 4		Sea–air ΔpCO ₂ (µatm) Transect 4		Sea surface <i>p</i> CO ₂ (µatm)		
							Transect 4		
			Stations 27-32 ^a	Station 33 ^b	Stations 27-32 ^a	Station 33 ^b	Stations 27-32 ^a	Station 33 ^b	
800	20	420	-25 ± 3	-25 ± 1	-60 ± 6	-59 ± 3	330 ± 6	331 ± 3	
1000	15	450	-23 ± 2	-23 ± 1	-55 ± 5	-53 ± 3	335 ± 5	337 ± 3	
1200	10	470	-22 ± 2	-22 ± 1	-53 ± 5	-52 ± 3	337 ± 5	338 ± 3	
			Transect 5 Transect 5 Stations 35–38 ^b Stations 35–38 ^b		ct 5	Transect 5			
					Stations 35–38 ^b		Stations 35–38 ^b		
800	20	540	-23 ± 3		-53 ± 8		337 ± 8		
1000	15	585	-20 ± 3		-48 ± 8		342 ± 8		
1200	10	610	$-19 \pm$	-19 ± 3		-46 ± 8		344 ± 8	
			Transect 6		Transect 6		Transect 6		
			Stations 45–49 ^b		Stations 45-49 ^b		Stations 45-49b		
800	20	500	-25 ± 3		-57 ± 6		333 ± 6		
1000	15	540	-23 ± 3		-53 ± 6		337 ± 6		
1200	10	570	-22 ± 3		-51 ± 6		339 ± 6		

^a Data for these stations were obtained from waters immediately below the surface buoyant layer. ^b Data for these stations were obtained from the surface mixed layer.

Table 2. Values of $\Delta DIC - 6.6 \Delta NO_3$, sea-air ΔpCO_2 and sea surface pCO_2 estimated with the deep water end member from different depths.

Depth of the deep water end member (m)	$\Delta DIC - 6.6 \Delta NO_3$ ($\mu mol kg^{-1}$) Transect 4		Sea–air ∆ (µatn		Sea surface <i>p</i> CO ₂ (µatm)			
			Transect 4		Transect 4			
	Stations 27–32 ^a	Station 33 ^b	Stations 27–32 ^a	Station 33 ^b	Stations 27–32 ^a	Station 33 ^b		
\sim 130	-23 ± 2	-19 ± 1	-56 ± 5	-45 ± 3	334 ± 5	345 ± 3		
~ 150	-23 ± 2	-22 ± 1	-55 ± 5	-52 ± 3	335 ± 5	338 ± 3		
~ 175	-23 ± 2	-23 ± 1	-55 ± 5	-53 ± 3	335 ± 5	337 ± 3		
~ 200	-23 ± 2	-24 ± 1	-55 ± 5	-56 ± 3	335 ± 5	334 ± 3		
	Transect 5 Stations 35–38 ^b		Transe	Transect 5		Transect 5		
			Stations 35–38 ^b		Stations 35–38 ^b			
~ 130	-21 ± 3		-51 ± 8		339 ± 8			
~ 150	-20 ± 3		-46	-46 ± 8		344 ± 8		
~ 175	-20 ± 3		-48	-48 ± 8		342 ± 8		
~ 200	-17 ± 3		-40 =	⊢ 8	350 ± 8			
	Transect 6 Stations 45–49 ^b		Transect 6		Transect 6			
			Stations 45–49 ^b		Stations 45–49 ^b			
~ 130	-20 ± 3		-46 ± 6		344 ± 6			
~ 150	-22 ± 3		-51 ± 6		339 ± 6			
~ 175	-23 ±	= 3	-53 =	-53 ± 6		337 ± 6		
~ 200	-21 ± 3		$-50 \pm$	Ė6	340 ± 6			

^a Data for these stations were obtained from waters immediately below the surface buoyant layer. ^b Data for these stations were obtained from the surface mixed layer.

important in resolving the CO_2 fluxes. We thus conducted a sensitivity analysis for these two sets of variables for the CO_2 sink zones off Oregon and northern California where our diagnostic approach worked well (i.e., waters immediately below the surface buoyant layer at stations 27–32 as well as waters in the surface mixed layer at stations 35–38 on transect 4, waters in the surface mixed layer at stations 45–49 on transect 6).

3.4.1 The combined freshwater end member

While the values of ~ 1000 and $\sim 15 \,\mu\text{mol}\,\text{kg}^{-1}$ were selected for TAlk/DIC and NO₃ in the mainstream of the CR, the field-observed TAlk and NO3 vary within a range of $\sim 800-1200$ (Evans et al., 2013) and $\sim 10-20 \,\mu mol \, kg^{-1}$ (http://www.stccmop.org/datamart/) in spring/early summer. We thus took the values of ~ 800 and $\sim 1200 \,\mu\text{mol}\,\text{kg}^{-1}$ as the lower and upper limit of the TAlk and DIC end members and those of ~ 10 and $\sim 20 \,\mu\text{mol}\,\text{kg}^{-1}$ as the lower and upper limit of the NO₃ end member in the CR to test the diagnostic approach. Following the same calculation of the combined freshwater end member (X^{eff} in Eq. 3), the lower and upper limit of DIC^{eff} was estimated to be ~420 and $\sim\!470\,\mu mol\,kg^{-1}$ for waters immediately below the surface buoyant layer at stations 27-32 and in the surface mixed layer at station 33 on transect 4. Those values were \sim 540 and $\sim 610\,\mu mol\,kg^{-1}$ for waters in the surface mixed layer at stations 35–38 on transect 5, and \sim 500 and \sim 570 μ mol kg⁻¹ in the surface mixed layer at stations 45-49 on transect 6. The NO_2^{eff} in every scenario was still zero.

The newly diagnosed $\Delta \text{DIC}-6.6\Delta \text{NO}_3$ and sea-air $\Delta p\text{CO}_2$ on transects 4, 5 and 6 displayed no difference with those with the initial TAlk and DIC of ~ 1000 µmol kg⁻¹ and NO₃ of ~ 15 µmol kg⁻¹ in the CR (Table 1), while all estimated sea surface $p\text{CO}_2$ values were within error (1 standard deviation) compared to the field measurements. Although the TAlk and DIC end members had large variations of up to ~ 400 µmol kg⁻¹ while NO₃ varied within ~ 10 µmol kg⁻¹ in the mainstream of the CR, the corresponding range of DIC^{eff} contributing to waters beyond the CR plume significantly decreased by approximately 1 order of magnitude, implying minor influence of its variations on our diagnosis of the CO₂ fluxes.

3.4.2 The deep water end member

We selected values at ~ 175 m as the deep water end member based on the TAlk–Sal relationship, whereas this end member depth might not be spatially stable in a highly dynamic upwelling system. Previous studies also show that the upwelling source water on the Oregon–California shelf can vary between 150 and 200 m (e.g., Hales et al., 2005; Feely et al., 2008). We thus tested the diagnostic approach with values at three other depths of ~ 130 , ~ 150 and ~ 200 m. On transects 4 and 6, the newly estimated $\Delta DIC-6.6\Delta NO_3$, sea-air ΔpCO_2 and sea surface pCO_2 using end-member values at both ~ 150 and ~ 200 m agreed well with those using end-member values at ~ 175 m, while the three variables were slightly higher using end-member values at ~ 130 m (Table 2). On transect 5, the newly estimated $\Delta DIC-6.6\Delta NO_3$, sea-air ΔpCO_2 and sea surface pCO_2 using end-member values at both ~ 130 and ~ 150 m agreed well with those using end-member values at ~ 175 m, while the three variables were slightly higher using end-member values at ~ 175 m, while the three variables were slightly higher using end-member values at ~ 100 m (Table 2).

3.4.3 The C/N uptake ratio

In a given oceanic setting, the real C/N uptake ratio during organic carbon production can be different from the Redfield stoichiometry of \sim 6.6 (Redfield et al., 1963). For instance, higher ratios estimated from the DIC-NO₃ relationship are observed in both coastal waters and open ocean sites, possibly resulting from excess DIC uptake via the production of dissolved organic carbon (Sambrotto et al., 1993; Ianson et al., 2003). However, since the precise estimation of the C/N uptake ratio (via, for example, in situ incubation experiments) is still problematic, such data are currently scarce over the world's oceans, and the empirical stoichiometry is routinely applied into field studies investigating the dynamics and coupling of carbon and nutrients (e.g., Chen et al., 2008; Fassbender et al., 2011). Fassbender et al. (2011) applied another empirical C/N uptake ratio of 7.3 (approximately 117/16; Anderson and Sarmiento, 1994) into the same data set as this study. We thus performed a simple sensitivity analysis using this alternative value of 7.3, which implies excess DIC uptake relative to NO₃.

Since $\Delta DIC - 7.3\Delta NO_3$ values were obviously smaller than $\Delta DIC - 6.6\Delta NO_3$ ones, the new sea-air ΔpCO_2 values were halved (Table 3). Correspondingly, the newly estimated sea surface pCO_2 values on transects 4, 5 and 6 were $\sim 35-45$ µatm lower than the estimation using the Redfield ratio, which were however consistent with the field measurements. Given that the Redfield ratio also works in our OceMar case studies of the SCS and the CS (Dai et al., 2013), we contend that this classic ratio could be preferentially employed if the field-observed elemental stoichiometry is not available. Moreover, as Martz et al. (2014) pointed out, "treating the Redfield ratios as global or regional constants may be acceptable in the context of interpreting snapshots of the water column captured in shipboard bottle data".

The above notion was also supported by examining the slope of the linear regression between DIC and NO₃ normalized to a constant salinity in the surface water or in the surface mixed layer, which provides an alternative to the C/N uptake ratio associated with organic carbon production (Sambrotto et al., 1993; Wong et al., 2002; Ianson et al., 2003). Given a nonzero combined freshwater end member, we adopted in this study an approach of regional

Table 3. Values of sea–air ΔpCO_2 and sea surface pCO_2 estimated with different $\Delta DIC - x \Delta NO_3$. The variable x denotes the C/N uptake ratio during organic carbon production. T and S represent transect and station(s).

C/N uptake ratio	$\Delta \text{ DIC}-x\Delta \text{NO}_3 \\ (\mu \text{mol kg}^{-1})$			Sea–air $\Delta p CO_2$ (µatm)				Sea surface <i>p</i> CO ₂ (µatm)				
	<i>T</i> 4		Τ5	Τ6	<i>T</i> 4		Τ5	Τ6	<i>T</i> 4		Τ5	Τ6
	S27-32 ^a	\$33 ^b	\$35-38 ^b	S45-49 ^b	S27-32 ^a	\$33 ^b	\$35-38 ^b	S45-49 ^b	S27-32 ^a	\$33 ^b	\$35-38 ^b	S45-49 ^b
6.6 ^c	-23 ± 2	-23 ± 1	-20 ± 3	-23 ± 3	-55 ± 5	-53 ± 3	-48 ± 8	-53 ± 6	335 ± 5	337 ± 3	342 ± 8	337 ± 6
7.3 ^d	-42 ± 3	-42 ± 1	-39 ± 3	-42 ± 3	-101 ± 6	-100 ± 3	-92 ± 8	-97 ± 6	289 ± 6	290 ± 3	298 ± 8	293 ± 6

^a Data for these stations were obtained from waters immediately below the surface buoyant layer. ^b Data for these stations were obtained from the surface mixed layer. ^c 6.6 is the Redfield C/N uptake ratio (approximately 106/16; Redfield et al., 1963). ^d 7.3 is the more recent evaluation of the C/N uptake ratio (approximately 117/16; Anderson and Sarmiento, 1994).

normalization (Friis et al., 2003; Cao et al., 2011) as:

$$nX = \frac{X^{\text{meas}} - X^{\text{eff}}}{\text{Sal}^{\text{meas}}} \cdot \text{Sal}^{\text{aver}} + X^{\text{eff}}.$$
(8)

Here, nX and X^{meas} are salinity-normalized and fieldmeasured values for DIC and NO₃. Sal^{meas} is the CTDmeasured salinity. Sal^{aver} is the average salinity value of ~33.0 in these CO₂ sink zones, which was selected as the constant salinity. X^{eff} is the same as that in Eq. (3), denoting the effective concentration of DIC or NO₃ sourced from the freshwater input to various zones off Oregon and northern California. While the NO₃^{eff} in any combined freshwater end member was zero, the DIC^{eff} was ~450 µmol kg⁻¹ for waters immediately below the surface buoyant layer at stations 27–32 and waters in the surface mixed layer at stations 25 and 33 on transect 4, ~585 µmol kg⁻¹ for waters in the surface mixed layer at stations 35–38 on transect 5, and ~540 µmol kg⁻¹ for waters in the surface mixed layer at stations 45–49 on transect 6.

As shown in Fig. 6, our analysis with all data from the CO_2 sink zones along the three transects revealed a slope of 6.70 ± 0.37 . This value was within error comparable to that of 6.6, suggesting that using the Redfield ratio in our diagnostic approach should be in order. On the other hand, we contend that scrutinizing the in situ C/N uptake ratio via relatively direct observations is imperative for better understanding the issue regarding the possible departure from the Redfield ratio.

4 Concluding remarks

The semi-analytical diagnostic approach of mass balance that couples physical transport and biogeochemical alterations was well applied to the CO_2 sink zones off Oregon and northern California in spring/early summer 2007, extending from the outer shelf to the open basin. In these zones with the absence of any significant influence of the CR plume, the source of DIC was largely from deep waters in the subtropical gyre of the eNP, and the ultimate CO_2 sink nature was determined by the higher nutrient consumption than DIC in the upper waters. On the other hand, the estimated CO_2 flux was

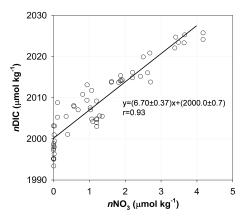


Figure 6. Salinity-normalized DIC (*n*DIC) versus salinitynormalized NO₃ (*n*NO₃) in the CO₂ sink zones off Oregon and northern California in spring/early summer 2007, which included waters immediately below the surface buoyant layer at stations 27-32 as well as waters in the surface mixed layer at station 33 on transect 4, waters in the surface mixed layer at stations 35-38 on transect 5 and waters in the surface mixed layer at stations 45-49on transect 6.

opposite the field observations in the nearshore upwelling zone along the Oregon-California coast, which behaved like a typical OceMar system in terms of its mixing process. This discrepancy was very likely due to minor biological responses during the intensified upwelling period, making our mass balance approach based on the coupled physical biogeochemistry invalid. This suggested that the applicability of the proposed semi-analytical diagnostic approach is limited to steady state systems with comparable timescales of water mass mixing and biogeochemical reactions. In a physical mixing prevailing regime, resolving the CO₂ fluxes could instead be simplified without considering the biological consumption of DIC and nutrients. Further work is however required to better understand the carbon and nutrient dynamics and the timing of the physical and biological processes associated with coastal upwelling.

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